**ANNAMALAI UNIVERSITY**

**(Affiliated Colleges)**

**204 - B.Sc. Chemistry**

Programme Structure and Scheme of Examination (under CBCS)

(Applicable to the candidates admitted from the academic year 2023 -2024 onwards)

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| --- | --- | --- | --- | --- | --- | --- | --- |
| Part | Course Code | Study Components & Course Title | Credit | Hours | Maximum Marks | | |
| CIA | ESE | Total |
|  |  | SEMESTER – I |  |  |  |  |  |
| I | 23UTAML11/  23UHINL11/ 23UFREL11 | Language – I:  பொதுதமிழ் – I/  Hindi – I/  French – I | 3 | 6 | 25 | 75 | 100 |
| II | 23UENGL12 | General English – I | 3 | 6 | 25 | 75 | 100 |
| III | 23UCHEC13 | Core – I: General Chemistry-I | 5 | 5 | 25 | 75 | 100 |
| 23UCHEP14 | Core –II : Practical - I  Quantitative Inorganic Estimation (Titrimetry) and Inorganic Preparations | 5 | 4 | 25 | 75 | 100 |
| 23UMATE15  23UBOTE15  23UZOOE15  23UBIOE15 | Elective - I (Generic / Discipline Specific)  Mathematics-1 (or)  Botany – I (or)  Zoology – I (or)  Biochemistry – I | 3/2 | 5/3 | 25 | 75 | 100 |
| 23UBOTEP1  23UZOOEP1  23UBIOEP1 | Botany Practical – I (or)  Zoology Practical – I (or)  Biochemistry Practical–I | 1 | 2 | 25 | 75 | 100 |
| IV | 23UTAMB16  23UTAMA16 | Skill Enhancement Course – 1 (NME-I) /\*  Basic Tamil – I /  Advanced Tamil - I | 2 | 2 | 25 | 75 | 100 |
| 23UCHEF17 | Foundation Course : Introductory Chemistry | 2 | 2 | 25 | 75 | 100 |
|  |  | Total | 23 | 30 |  |  | 700/800 |
|  |  | SEMESTER – II |  |  |  |  |  |
| I | 23UTAML21/ 23UHINL21/ 23UFREL21 | Language – II:  பொது தமிழ் –II/  Hindi – II/  French –II | 3 | 6 | 25 | 75 | 100 |
| II | 23UENGL22 | General English – II | 3 | 6 | 25 | 75 | 100 |
| III | 23UCHEC23 | Core – III: General Chemistry-II | 5 | 5 | 25 | 75 | 100 |
| 23UCHEP24 | Core –IV: Practical– II  Qualitative Organic Analysis and Preparation of Organic Compounds | 5 | 4 | 25 | 75 | 100 |
| 23UMATE25  23UBOTE25  23UZOOE25  23UBIOE25 | Elective - II (Generic / Discipline Specific)  Mathematics-II (or)  Botany-II (or)  Zoology-II (or)  Biochemistry -II | 3/2 | 5/3 | 25 | 75 | 100 |
| 23UBOTEP2  23UZOOEP2  23UBIOEP2 | Botany Practical – II (or)  Zoology Practical – II (or)  Biochemistry Practical –II | 1 | 2 | 25 | 75 | 100 |
| IV | 23UTAMB26  23UTAMA26 | Skill Enhancement Course – 2 (NME-II) /\*  Basic Tamil – II /  Advanced Tamil - II | 2 | 2 | 25 | 75 | 100 |
| 23USECG27 | Skill Enhancement Course – 3  Internet and its Applications  (Common Paper) | 2 | 2 | 25 | 75 | 100 |
|  |  | Total | 23 | 30 |  |  | 700/800 |

Non-major (NME) Electives offered to other Departments

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| --- | --- | --- | --- | --- | --- | --- | --- |
| IV | 23UCHEN16 | Role of Chemistry in Daily Life | 2 | 2 | 25 | 75 | 100 |
| 23UCHEN26 | Dairy Chemistry | 2 | 2 | 25 | 75 | 100 |

\* PART-IV: NME / Basic Tamil / Advanced Tamil (Any one)

Students who have not studied Tamil upto 12th Standardand have taken any Language other than Tamil in Part-I, must choose Basic Tamil-I in First Semester & Basic Tamil-II in Second Semester.

Students who have studied Tamil upto 10th & 12th Standardand have taken any Language other than Tamil in Part-I, must choose Advanced Tamil-I in First Semester and Advanced Tamil-II in Second Semester.

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| **SEMESTER: I**  **PART: III**  **Core – I** | **23UCHEC13**  **GENERAL CHEMISTRY-I** | **Credit: 5**  **Hours: 5** |

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| Objectives of the course | The course aims at giving an overall view of the   * various atomic models and atomic structure * wave particle duality of matter * periodic table, periodicity in properties and its application in explaining the chemical behaviour * nature of chemical bonding, and * fundamental concepts of organic chemistry |
| Course Outline | UNIT I  Atomic structure and Periodic trends  History of atom (J.J.Thomson, Rutherford); Moseley’s Experiment and Atomic number, Atomic Spectra; Black-Body Radiation and Planck’s quantum theory Bohr's model of atom;The Franck-Hertz Experiment; Interpretation of Hspectrum; Photoelectric effect, Compton effect; Dual nature of Matter- DeBroglie wavelength-Davisson and Germer experiment Heisenberg’s Uncertainty Principle; Electronic Configuration of Atoms and ions- Hund’s rule, Pauli’exclusion principle and Aufbau principle;  Numerical problems involving the core concepts. |
| Unit II  Introduction to Quantum mechanics  Classical mechanics, Wave mechanical model of atom, distinction between a Bohr orbit and orbital; Postulates of quantum mechanics; probability interpretation of wavefunctions, Formulation of Schrodinger wave equation - Probability and electron density-visualizing the orbitals -Probability density and significance of Ψ and Ψ2.  Modern Periodic Table  Cause of periodicity; Features of the periodic table; classification of elements Periodic trends for atomic size- Atomic radii, Ionic, crystal and Covalent radii; ionization energy, electron affinity, electronegativity-electronegativity scales, applications of electronegativity.  Problems involving the core concepts |

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|  | UNIT-III: Structure and bonding - I  Ionic bond  Lewis dot structure of ionic compounds; properties of ionic compounds; Energy involved in ionic compounds; Born Haber cycle – lattice energies, Madelung constant; relative effect of lattice energy and solvation energy; Ion polarisation – polarising power and polarizability; Fajans’ rules - effects of polarisation on properties of compounds; problems involving the core concepts.  Covalent bond  Shapes of orbitals, overlap of orbitals – σ and Π bonds; directed valency hybridization; VSEPR theory - shapes of molecules of the type AB2, AB3, AB4, AB5, AB6 and AB7  Partial ionic character of covalent bond-dipole moment, application to molecules of the type A2, AB, AB2, AB3, AB4; percentage ionic character- numerical problems based on calculation of percentage ionic character. |
| UNIT-IV: Structure and bonding - II  VB theory – application to hydrogen molecule; concept of resonance - resonance structures of some inorganic species – CO2, NO2, CO32-, NO3- ; limitations of VBT; MO theory - bonding, antibonding and nonbonding orbitals, bond order; MO diagrams of H2, C2, O2, O2+, O2-, O22-N2, NO, HF, CO; magnetic characteristics, comparison of VB and MO theories.  Coordinate bond: Definition, Formation of BF3, NH3, NH4+, H3O+ properties  Metallic bond-electron sea model, VB model; Band theory-mechanism of conduction in solids; conductors, insulator, semiconductor – types, applications of semiconductors  Weak Chemical Forces - Vander Waals forces, ion-dipole forces, dipole-dipole interactions, induced dipole interactions, Instantaneous dipole-induced dipole interactions. Repulsive forces; Hydrogen bonding – Types, special properties of water, ice, stability of DNA; Effects of chemical force, melting and boiling points. |
| UNIT-V:  Basic concepts in Organic Chemistry and Electronic effects    Types of bond cleavage – heterolytic and homolytic; arrow pushing in organic reactions; reagents and substrates; types of reagents - electrophiles, nucleophiles, free radicals; reaction intermediates – carbanions, carbocations, carbenes, arynes and nitrynes.  Inductive effect - reactivity of alkyl halides, acidity of halo acids, basicity of amines; inductomeric and electromeric effects.  Resonance – resonance energy, conditions for resonance - acidity of phenols, basicity of aromatic amines, stability of carbonium ions, carbanions and free |

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|  | radicals, reactivity of vinyl chloride, dipole moment of vinyl chloride and nitrobenzene, bond lengths; steric inhibition to resonance.  Hyperconjugation - stability of alkenes, bond length, orienting effect of methyl group, dipole moment of aldehydes and nitromethane  Types of organic reactions- addition, substitution, elimination and rearrangements |
| Extended  Professional Component (is a part of internal  component only, Not to be included in the external examination question paper) | Questions related to the above topics, from various competitive examinations  UPSC/JAM /TNPSC and others to be solved  (To be discussed during the Tutorial hours) |
| Skills acquired from this course | Knowledge, Problem solving, Analytical ability, Professional Competency, Professional Communication and Transferable skills. |
| Recommended  Text | 1. Madan, R. D. and Sathya Prakash, Modern Inorganic Chemistry, 2nded.; S. Chand and Company: New Delhi, 2003. 2. Rao, C.N. R. University General Chemistry, Macmillan Publication: New Delhi, 2000. 3. Puri, B. R. and Sharma, L. R. Principles of Physical Chemistry, 38thed.;Vishal Publishing Company: Jalandhar, 2002. 4. Bruce, P. Y. and PrasadK. J. R. Essential Organic Chemistry, Pearson Education: New Delhi, 2008. 5. Dash UN, Dharmarha OP, Soni P.L. Textbook of Physical Chemistry,   Sultan Chand & Sons: New Delhi,2016 |
| Reference  Books | 1. Maron, S. H. and Prutton C. P. Principles of Physical Chemistry,4thed.; The Macmillan Company: Newyork,1972. 2. Lee, J. D. Concise Inorganic Chemistry, 4th ed.; ELBS William Heinemann: London,1991. 3. Gurudeep Raj, Advanced Inorganic Chemistry, 26thed.; Goel Publishing House: Meerut, 2001. 4. Atkins, P.W. & Paula, J. Physical Chemistry, 10th ed.; Oxford University Press:New York, 2014. 5. Huheey, J. E. Inorganic Chemistry: Principles of Structure and Reactivity, 4th ed .; Addison, Wesley Publishing Company: India,1993. |
| Website and e-learning source | 1. https://onlinecourses.nptel.ac.in 2. http://www.mikeblaber.org/oldwine/chm1045/notes\_m.htm 3. http://www.ias.ac.in/initiat/sci\_ed/resources/chemistry/Inorganic.html 4. https://swayam.gov.in/course/64-atomic-structure-and-chemical-bonding 5) https://www.chemtube3d.com/ |
| Course Learning Outcomes (for Mapping with POs and PSOs)  On completion of the course the students should be able to  CO1: explain the atomic structure, wave particle duality of matter, periodic properties bonding, and properties of compounds.  CO2: classify the elements in the periodic table, types of bonds, reaction intermediates electronic effects in organic compounds, types of reagents.  CO3: apply the theories of atomic structure, bonding, to calculate energy of a spectral transition, Δx, Δp electronegativity, percentage ionic character and bond order.  CO4: evaluate the relationship existing between electronic configuration, bonding, geometry of molecules and reactions; structure reactivity and electronic effects  CO5: construct MO diagrams, predict trends in periodic properties, assess the properties of elements, and explain hybridization in molecules, nature of H – bonding and organic reaction mechanisms. | |

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|  | PO1 | PO2 | PO3 | PO4 | PO5 | PO6 | PO7 | PO8 | PO9 | PO10 |
| CO1 | S | S | S | S | S | S | S | M | S | M |
| CO2 | M | S | S | S | M | S | S | M | M | M |
| CO3 | S | S | S | M | S | S | S | M | S | M |
| CO4 | S | S | S | S | S | S | S | M | M | M |
| CO5 | S | M | S | S | S | S | S | M | M | S |

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| --- | --- | --- | --- | --- | --- |
| CO /PSO | PSO1 | PSO2 | PSO3 | PSO4 | PSO5 |
| CO1 | 3 | 3 | 3 | 3 | 3 |
| CO2 | 3 | 3 | 3 | 3 | 3 |
| CO3 | 3 | 3 | 3 | 3 | 3 |
| CO4 | 3 | 3 | 3 | 3 | 3 |
| CO5 | 3 | 3 | 3 | 3 | 3 |
| Weightage | 15 | 15 | 15 | 15 | 15 |
| Weighted percentage of  Course Contribution to Pos | 3.0 | 3.0 | 3.0 | 3.0 | 3.0 |

Level of Correlation between PSO’s and CO’

|  |  |  |
| --- | --- | --- |
| **SEMESTER: I**  **PART: III**  **Core – II** | **23UCHEP14**  **Quantitative Inorganic Estimation (titrimetry) and Inorganic Preparations** | **Credit: 5**  **Hours: 4** |

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| Objectives of the course | This course aims at providing knowledge on   * laboratory safety * handling glasswares * Quantitative estimation * preparation of inorganic compounds * To develop the skill in finding out the end points of various types of indicators |
| Course Outline | Unit I  Chemical Laboratory Safety in Academic Institutions  Introduction - importance of safety education for students, common laboratory hazards, assessment and minimization of the risk of the hazards, prepare for emergencies from uncontrolled hazards; concept of MSDS; importance and care of PPE; proper use and operation of chemical hoods and ventilation system; fire extinguishers-types and uses of fire extinguishers, demonstration of operation; chemical waste and safe disposal.  Common Apparatus Used in Quantitative Estimation (Volumetric)  Description and use of burette, pipette, standard flask, measuring cylinder, conical flask, beaker, funnel, dropper, clamp, stand, wash bottle, watch glass, wire gauge and tripod stand.  Principle of Quantitative Estimation (Volumetric)    Equivalent weight of an acid, base, salt, reducing agent, oxidizing agent; concept of mole, molality, molarity, normality; primary and secondary standards, preparation of standard solutions; theories of acid-base, redox, complexometric, iodimetric and iodometric titrations; indicators – types, theory of acid–base, redox, metal ion and adsorption indicators, choice of indicators. |
|  | Dichrometry  Estimation of ferric alum using standard dichromate (external indicator) Estimation of ferric alum using standard dichromate (internal indicator)    Iodometry  Estimation of copper in copper sulphate using standard dichromate    Argentimetry  Estimation of chloride in barium chloride using standard sodium chloride/ Estimation of chloride in sodium chloride (Volhard’s method)  Complexometry  Estimation of hardness of water using EDTA  Estimation of iron in iron tablets  Estimation of ascorbic acid |
| Preparation of Inorganic compounds- Potash alum  Tetraammine copper (II) sulphate  Hexamminecobalt (III) chloride  Mohr’s Salt |
| Skills acquired from this course | Knowledge, Problem solving, Analytical ability, Professional Competency, Professional Communication and Transferable skills. |
| Recommended  Text | Reference Books:  1.Venkateswaran, V.;Veeraswamy, R.;Kulandivelu, A.R. Basic Principles of Practical Chemistry,2nd ed.; Sultan Chand &Sons: New Delhi, 1997.  2.Nad, A. K.; Mahapatra, B.; Ghoshal, A.; An advanced course in Practical Chemistry, 3rd ed.; New Central Book Agency: Kolkata, 2007. |
| Reference  Books | 1. Mendham, J.; Denney, R. C.; Barnes, J. D.; Thomas, M.; Sivasankar, B.; Vogel’s Textbook of Quantitative Chemical Analysis, 6th ed.; Pearson Education Ltd: New Delhi, 2000. |
| Website and e-learning source | Web References:  1) http://www.federica.unina.it/agraria/analytical-chemistry/volumetricanalysis  2)https://chemdictionary.org/titration-indicator/ |
| Course Learning Outcomes (for Mapping with POs and PSOs)  On successful completion of the course the students should be able to  CO1: explain the basic principles involved in titrimetric analysis and inorganic preparations.  CO2: compare the methodologies of different titrimetric analysis.  CO3: calculate the concentrations of unknown solutions in different ways and develop the skill to estimate the amount of a substance present in a given solution.  CO4.Indetify the end point of various titrations  CO5 acquire knowledge on the systematic analysis of Mixture of salts., identify the cations and anions in the unknown substance.  CO5: handle the common apparatus used in volumetric estimation. | |

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|  | PO1 | PO2 | PO3 | PO4 | PO5 | PO6 | PO7 | PO8 | PO9 | PO10 |
| CO1 | S | S | S | S | S | S | S | M | S | M |
| CO2 | M | S | S | S | M | S | S | M | M | M |
| CO3 | S | S | S | M | S | S | S | M | S | M |
| CO4 | S | S | S | S | S | S | S | M | M | M |
| CO5 | S | S | S | M | S | S | S | M | S | M |

CO-PO Mapping (Course Articulation Matrix)

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| --- | --- | --- | --- | --- | --- |
| CO /PSO | PSO1 | PSO2 | PSO3 | PSO4 | PSO5 |
| CO1 | 3 | 3 | 3 | 3 | 3 |
| CO2 | 3 | 3 | 3 | 3 | 3 |
| CO3 | 3 | 3 | 3 | 3 | 3 |
| CO4 | 3 | 3 | 3 | 3 | 3 |
| CO5 | 3 | 3 | 3 | 3 | 3 |
| Weightage | 15 | 15 | 15 | 15 | 15 |
| Weighted percentage of  Course Contribution to  Pos | 3.0 | 3.0 | 3.0 | 3.0 | 3.0 |

Note: Scheme for Practical Evaluation

Inorganic Preparation:15 marks

Record:10 marks

Procedure-10 marks

Volumetric Estimation – 40 marks

Results

< 2% - 40 marks

2-3% - 30 marks

3-4% - 20 marks

> 4% - 10 marks

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| **SEMESTER: I**  **PART: III** | **23UMATE15**  **MATHEMATICS – I** | **Credit: 3**  **Hours: 5** |

**UNIT-I: SOLUTIONS OF TRANSCENDENTAL AND ALGEBRAIC EQUATIONS**

Iteration method**,** Bisection method, Newton’s method - Regula Falsi method, (without proof) (Simple problems only)

**Unit-II: SOLUTIONS OF SIMULTANEOUS EQUATIONS**

Gauss Elimination method- Gauss Jordan method-Gauss Seidel Iterative method-Gauss Jacobi method (Restricted to three variables only) (Simple problems only)

**UNIT-III: MATRICES**

Characteristic equation of a square matrix– Eigen values and eigen vectors – Cayley – Hamilton theorem [without proof] – Verification and computation of inverse matrix.

**UNIT-IV: DIFFERENTIAL CALCULUS**

n-th derivatives – Leibnitz theorem [without proof] and applications – Jacobians– Curvature and radius of curvature in Cartesian co-ordinates

**UNIT-V: APPLICATION OF INTEGRATION**

Evaluation of double – Simple applications to area,

**TEXT BOOKS:**

1. A.Singaravelu “Numerical Methods”Meenakshi Publications

Unit-I: Chapter 2

Unit-II: Chapter 2

1. P. Duraipandian and Dr. S. Udayabaskaran. 1997, “Allied Mathematics” , Vol I & II. Chennai: Muhil Publishers.

Unit-III: Sec(4.5, 4.5.1 to 4.5.3)Vol I

Unit-IV: Sec(1.1.1, 1.1.2, 1.2, 1.4.3)vol II

Unit-V: Chap:3(3.2, 3.4, 3.4.1) vol II

**REFERENCE BOOKS:**

1. P. Balasubramanian and K. G. Subramanian. 1997, “Ancillary Mathematics”, Vol I & II. New Delhi: Tata McGraw Hill.
2. S.P.Rajagopalan and R.Sattanathan(2005), “Allied Mathematics”, Vol I & II. New Delhi: Vikas Publications.
3. P. R. Vittal (2003), “Allied Mathematics”,Chennai: Marghan Publications.

**Course Outcomes:**

On successful completion of the course, the students will be able to

**CO1**: Attain knowledge on finding Approximate root for polynomial equations using Numerical methods.

**CO2**: Develop the skills of finding solutions of Simultaneous Linear equations.

**CO3**: Adopt techniques in solving problems involving Matrices

**CO4**: Provide skills on finding curvature and radius of curvature in Cartesian and polar co-ordinates.

**CO5**:Understand the applications of double and Triple integration in real life situation.

**Outcome Mapping:**

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| --- | --- | --- | --- | --- | --- |
| CO / PO | PO1 | PO2 | PO3 | PO4 | PO5 |
| CO1 | 3 | 3 | 3 | 2 | 2 |
| CO2 | 3 | 2 | 3 | 3 | 2 |
| CO3 | 3 | 3 | 3 | 3 | 2 |
| CO4 | 3 | 3 | 3 | 3 | 2 |
| CO5 | 2 | 3 | 3 | 3 | 2 |

**1-Low 2-Moderate 3- High**

|  |  |  |
| --- | --- | --- |
| **SEMESTER: I**  **PART: III** | **23UBOTE15**  **Botany- I** | **Credit: 2**  **Hours: 3** |

**Learning Objective (LO):**

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| **LO1** | To study morphological and anatomical adaptations of plants of various habitats. |
| **LO2** | To demonstrate techniques of plant tissue culture. |
| **LP3** | To familiarize with the structure of DNA, RNA. |
| **LO4** | To carryout experiments related with plant physiology. |
| **LO5** | To perform biochemistry experiments. |

**Unit – 1: Algae:**

General characters of algae - Structure, reproduction and life cycle of the following genera - *Anabaena* and *Sargassum* and economic importance of algae.

**Unit – 2 : Fungi, Bacteria and Virus:**

General characters of fungi, structure, reproduction and life cycle of the following genera - *Penicillium* and *Agaricus* and economic importance of fungi.

Bacteria - general characters, structure and reproduction of *Escherichia coli* and economic importance of bacteria. Virus - general characters, structure of TMV, structure of bacteriophage.

**Unit – 3 : Bryophytes, Pteridophytes and Gymnosperms:**

General characters of Bryophytes, Structure and life cycle of *Funaria*.

General characters of Pteridophytes, Structure and life cycle of *Lycopodium*.

General characters of Gymnosperms, Structure and life cycle of *Cycas*.

**Unit – 4: Cell Biology:**

Prokaryotic and Eukaryotic cell- structure /organization. Cell organelles - ultra structure and function of chloroplast, mitochondria and nucleus. Cell division - mitosis and meiosis.

**Unit – 5: Genetics and Plant Biotechnology:**

Mendelism - Law of dominance, Law of segregation, Incomplete dominance. Law of independent assortment. Monohybrid and dihybrid cross - Test cross - Back cross. Plant tissue culture - *In vitro* culture methods. Plant tissue culture and its application in biotechnology.

**Course Outcomes (CO)**

At the end of the course, the student will be able to

|  |  |
| --- | --- |
| **CO1** | Increase the awareness and appreciation of human friendly  algae and their economic importance. |
| **CO2** | Develop an understanding of microbes and fungi and  appreciate their adaptive strategies. |
| **CO3** | Develop critical understanding on morphology, anatomy and  reproduction of Bryophytes, Pteridophytes and Gymnosperms. |
| **CO4** | Compare the structure and function of cells and explain the  development of cells. |
| **CO5** | Understand the core concepts and fundamentals of plant  biotechnology and genetic engineering. |

**Recommended Texts**

1. Singh, V., Pande, P.C and Jain, D.K. 2021. A Text Book of Botany. Rastogi Publications, Meerut.
2. Bhatnagar, S.P and Alok Moitra. 2020. Gymnosperms, New Age International (P) Ltd., Publishers, Bengaluru.
3. Sharma, O.P. 2017. Bryophyta, MacMillan India Ltd. Delhi.
4. Lee, R.E. 2008. Phycology, IV Edition, Cambridge University Press, New Delhi.
5. Rao, K., Krishnamurthy, K.V and Rao, G.S. 1979. Ancillary Botany,S.

Viswanathan Pvt. Ltd., Madras.

**Reference books:**

1. Parihar, N.S. 2012. An introduction to Embryophyta –Pteridophytes - Surjeet Publications, Delhi.
2. Alexopoulos, C.J. 2013. Introduction to Mycology. Willey Eastern Pvt. Ltd.
3. Vashishta, P.C. 2014. Botany for Degree Students Gymnosperms. Chand & Company Ltd, Delhi.
4. Coulter, M. Jhon, 2014. Morphology of Gymnosperms. Surjeet Publications, Delhi.
5. Vashishta, P.C. 2014. Botany for Degree Students Algae. 2014. Chand & Company Ltd, Delhi.
6. Parihar, N.S. 2013. An introduction to Embryophyta –Bryophytes -, Surjeet Publications, Delhi.

7. Pandey B.P. 1986, Text Book of Botany (College Botany) Vol I &II,

S.Chand and Co. New Delhi.

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| **SEMESTER: I**  **PART: III** | **23UZOOE15**  **Zoology – I** | **Credit: 2**  **Hours: 3** |

**Course Objectives**

The main objectives of this course are:

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| 1 | To acquire a basic knowledge of diversity and organization of Protozoa, Coelenterata, Helminthes and Annelida |
| 2 | To acquire a basic knowledge of diversity and organization of Arthropoda, Mollusca and Echinodermata |
| 3 | To comprehend the taxonomic position and diversity among Protochordata, Pisces and Amphibia |
| 4 | To comprehend the taxonomic position and diversity among Reptilia, Aves and Mammalia |
| 5 | To acquire detailed knowledge of selected invertebrate and chordate forms |

**Unit - I: Diversity of Invertebrates–**I

Principles of taxonomy. Criteria for classification–Symmetry and Coelom –Binomial nomenclature. Classification of Protozoa, Coelenterata, Helminthes and Annelida upto classes with two examples.

**Unit – II: Diversity of Invertebrates–II**

Classification of Arthropoda, Mollusca and Echinodermata upto class level with examples.

**Unit – III: Diversity of Chordates–I**

Classification of Prochordata, Pisces and Amphibia upto orders giving two examples.

**Unit – IV:**  **Diversity of Chordates–II**

Classification of Reptilia, Aves and Mammalia upto orders giving two examples.

**Unit –V : Animal organization**

Structure and organization of (i) Earthworm, (ii) Rabbit/Rat, (iii) Prawn/Fish

**Expected Course Outcomes**

On completion of this course, students will:

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| **1** | Recall the characteristic features invertebrates and chordates. |
| **2** | Classify invertebrates up to class level and chordates up to order level |
| **3** | Explain and discuss the structural and functional organisation of some invertebrates and chordates |
| **4** | Relate the adaptations and habits of animals to their habitat |
| **5** | Analyse the taxonomic position of animals. |

**Text Books (Latest Editions)**

* + - 1. Ekambaranatha Iyer, - Outlines of Zoology, Viswanathan Publication.

**References Books**

**(Latest editions, and the style as given below must be strictly adhered to)**

Ekambaranatha Iyar and T.N.Ananthakrishnian - A Manual of Zoology Invertebrata–VoI. I: Viswanathan Publishers.

Ekambaranatha Iyar and T.N. Ananthakrishnan, - A Manual of Zoology -Invertebrata–Vol. II: Viswanathan Publishers.

Ekambaranatha Iyar and T.N.Ananthakrishnan, - A Manual of Zoology: Chordata Viswanathan Publishers.

Jordan E.L. and P.S. Verma-Invertebrate Zoology, S. Chand & Co.

**Web Resources**

1. [www.sanctuaryasia.com](http://www.sanctuaryasia.com/)

2. [www.iaszoology.com](http://www.iaszoology.com/)

**Outcome Maping**

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| --- | --- | --- | --- | --- | --- | --- | --- | --- |
|  | **PO 1** | **PO 2** | **PO 3** | **PO 4** | **PO 5** | **PO 6** | **PO 7** | **PO 8** |
| **CO 1** | S |  |  |  |  |  |  |  |
| **CO 2** | M | S |  |  |  |  |  |  |
| **CO 3** |  |  |  | S |  | S |  |  |
| **CO 4** |  |  |  | S | S | M |  |  |
| **CO 5** |  |  | S |  |  |  |  | S |

**S-Strong M-Medium L-Low**

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| --- | --- | --- |
| **SEMESTER: I**  **PART: III** | **23UBIOE15**  **Biochemistry- I** | **Credit: 2**  **Hours: 3** |

**Learning objectives**

The objectives of this course are to

* Introduce the structure and classification of carbohydrates
* Comprehend the metabolism of carbohydrates
* Study the classification and properties of amino acids
* Elucidate the various levels of organization of Proteins
* Study functions and deficiency diseases of vitamins

**Module I:**Definition and classification of carbohydrates, linear and cyclic forms (Haworth projection) for glucose, fructose and mannose and disaccharides (maltose, lactose, sucrose).General properties of monosaccharides and disaccharides. Occurrence andsignificance of polysaccharides.12Hrs

**Module II**:Metabolism- Catabolism and Anabolism.Carbohydrate metabolism- Glycolysis, TCA cycle, HMP shunt and glycogen metabolism and energetics 12Hrs

**Module III:**Amino acids -Classifications, physical properties -amphoteric nature, isoelectric point and chemicalreactions of carboxyl ,amino and both groups. Amino acid metabolism- transamination, deamination and decarboxylation.12Hrs

**Module IV :**Proteins- classification - biological functions ,physical properties- ampholytes, iso electric point, salting in and salting out, denaturation, nature of peptide bond. Secondary structure, α-helix and β-pleated sheet, tertiary structure, various forces involved- quaternary structure. 12Hrs

**Module V:** Vitamins- Fat(A,D,E and K) and water soluble vitamins( B complex and C)- sources, RDA, biological functions and deficiency diseases12 Hrs

**Course Outcome**

|  |  |  |
| --- | --- | --- |
| **CO** | **On completion of this course, students will be able to** | **Programme**  **Outcome** |
| CO1 | Classify the structure of carbohydrates and its properties | PO1 |
| CO2 | Explain the metabolism of carbohydrates and its significance | PO1 |
| CO3 | Classify amino acids and its properties | PO1 |
| CO4 | Explain the classification and elucidate the different levels of structural organization of proteins | PO1 |
| CO5 | Identify the disease caused by the deficiency of vitamins | PO1 |

**Text Books**

1 Satyanarayan,U (2014) Biochemistry (4th ed), Arunabha Sen Books & Allied (P) Ltd, Kolkata.

2.Jain J.L.(2007) Fundamentals of Biochemistry,S.Chand publishers 311

**Reference books**

**1.** David L.Nelson and Michael M.Cox (2012) Lehninger Principles of Biochemistry (6th ed) W.H. Freeman.

**2.** Voet.D&Voet. J.G (2010) Biochemistry , (4th ed), John Wiley & Sons, Inc.

**3.** Lubert Stryer (2010) Biochemistry,(7th ed), W.H.Freeman

**4.** Satyanarayan,U (2014) Biochemistry (4th ed), Arunabha Sen Books & Allied (P) Ltd, Kolkata.

5.Jain J.L.(2007) Fundamentals of Biochemistry,S.Chand publishers 31

**Web sources**

1.onlinecourses.swayam2.ac.in/cec20\_bt12

2 onlinecourses.swayam2.ac.in/cec20\_bt19

**Mapping with Program Outcome**

|  |  |  |  |  |  |  |  |  |  |  |
| --- | --- | --- | --- | --- | --- | --- | --- | --- | --- | --- |
|  | **PO 1** | **PO 2** | **PO 3** | **PO 4** | **PO 5** | **PO 6** | **PSO1** | **PSO2** | **PSO3** | **PSO4** |
| **CO 1** | 3 |  |  |  |  |  | 3 |  |  | 3 |
| **CO 2** | 3 |  |  |  |  |  | 3 |  |  | 3 |
| **CO 3** | 3 |  |  |  |  |  | 3 |  |  | 3 |
| **CO 4** | 3 |  |  |  |  |  | 3 |  |  | 3 |
| **CO5** | 3 |  |  |  |  |  | 3 | 3 |  | 3 |

**S - Strong (3) M - Medium (2) L -Low(1)**

|  |  |  |
| --- | --- | --- |
| SEMESTER –I  PART – III | Elective  23UBOTEP1: Botany Practical I | CREDITS: 1  HOURS: 30 |

**Course Objectives**

1. To enhance information on the identification of each taxonomical group by developing the skill-based detection of the morphology and microstructure of microorganisms, algae, and fungi
2. To comprehend the fundamental concepts and methods used to identify Bryophytes, Pteridophytes and Gymnosperms through morphological changes and evolution, anatomy and reproduction.
3. To be familiar with the basic concepts and principles of cell biology.
4. Understanding of laws of inheritance, genetic basis of loci and alleles.
5. To learn about the principles and applications of Biotechnology

**EXPERIMENTS**

1. Make suitable micro preparation of the types prescribed in Algae, Fungi, Bryophytes, Pteridophytes and Gymnosperms.
2. Micro photographs of the cell organelles ultra structure.
3. Simple genetic problems.
4. Spotters - Algae, Fungi, Bryophytes, Pteridophytes, Gymnosperms Cell biology and Biotechnology.

**Bonafide record of practical work done should be submitted for the practical examination**

**Course outcomes:**

On completion of this course, the students will be able to:

1. To study the internal organization of algae .
2. To study the structure and organization of fungi, bacteria and viruses
3. Develop critical understanding on morphology, anatomy and reproduction of

Bryophytes, Pteridophytes and Gymnosperms.

1. To study the cell structure and function.
2. Understand the fundamental concepts of genetics and Biotechnology

**Recommended texts**

1. Sharma, O.P. 2017. Bryophyta, MacMillan India Ltd, New Delhi.

2. Sharma, O.P. 2012. Pteridophyta, Tata McGraw-Hills Ltd, New Delhi.

3. Subramaniam, N.S. 1996. Laboratory Manual of Plant Taxonomy. Vikas Publishing House Pvt. Ltd., New Delhi.

4. Benjamin, A. Pierce. 2012. Genetics- A conceptual Approach. W.H. Freeman and Company, New York, England.

5.Noggle G.R and G.J. Fritz. 2002. Introductory Plant Physiology. Prentice Hall of India, New Delhi.

**Reference books**

1. Strickberger, M.W. 2005. Genetics (III Ed). Prentice Hall, New Delhi, India.
2. Nancy Serediak and M. Huynh. 2011. Algae identification lab Guide. Accompanying manual to algae identification field guide, Ottawa Agriculture and Agri food Canada publisher.
3. Mohammed Gufran Khan, Shite Gatew and Bedilu Bekele. 2012. Practical manual for Bryophytes and Pteridophytes. Lambert Academic Publishing.
4. Aler Gingauz. 2001. Medicinal Chemistry. Oxford University Press & Wiley Publications.
5. Steward, F.C. 2012. Plant Physiology Academic Press, US

**Web Resources**

1. https://www.amazon.in/Practical-Manual-Pteridophyta-Rajan-Sundara/dp/8126106883
2. [https://www.google.co.in/books/edition/Gymnosperms/3YrT5E3Erm8C?hl=en&gbpv=1&dq=gy](https://www.google.co.in/books/edition/Gymnosperms/3YrT5E3Erm8C?hl=en&gbpv=1&dq=gymnosperms&printsec=frontcover) [mnosperms&printsec=frontcover](https://www.google.co.in/books/edition/Gymnosperms/3YrT5E3Erm8C?hl=en&gbpv=1&dq=gymnosperms&printsec=frontcover)
3. <https://www.amazon.in/Manual-Practical-Bryophyta-Suresh-Kumar/dp/B0072GNFX4>

OUTCOME MAPPING

|  |  |  |  |  |  |
| --- | --- | --- | --- | --- | --- |
|  | **PO1** | **PO2** | **PO3** | **PO4** | **PO5** |
| **CO1** | 2 | 3 | 1 | 3 | 2 |
| **CO2** | 2 | 3 | 2 | 2 | 3 |
| **CO3** | 2 | 1 | 3 | 2 | 3 |
| **CO4** | 1 | 3 | 3 | 2 | 2 |
| **CO5** | 2 | 2 | 3 | 1 | 3 |

**BOTANY PRACTICAL I**

Time : 3 Hours Max. Marks : 75

**PRACTICAL QUESTION PAPER**

1. Make suitable micro preparations of the given specimens A, B and C.

Submit the slides for valuation. Identify the specimens, draw diagrams and give reasons.

(Identification – 1, diagram – 2, Reasons – 2, Slide -2) (7 X 3) (21)

2. Make suitable micro preparations of the given specimens D.

Submit the slides for valuation. Identify the specimens, draw diagrams and give reasons.

(Identification – 1, diagram – 2, Reasons – 2, Slide -2) (08)

3. Identify the given electron micrograph –E, describe and draw diagrams

(Identification – 2, Diagram – 3, description – 3) (08)

4. Spotters – F, G, H, I, J, K and L.

(Identification – 1, diagram – 1, Reasons – 2) (7 X 4) (28)

Total = 65

Record = 10

\_\_\_\_\_\_\_\_\_\_\_\_

Grand Total = 75

\_\_\_\_\_\_\_\_\_\_\_\_

**BOTANY PRACTICAL I**

**KEY & SCHEME OF VALUATION**

1. A – Algae / Fungi : *Sargassum/Agaricus*

B – Bryophytes : *Funaria*

C – Pteridophytes : *Lycopodium*

(Identification – 1, diagram – 2, Reasons – 2, Slide -2) (7 X 3) (21)

2. Gymnopserms - D : *Cycas* – rachis and leaflet

(Identification – 1, diagram – 2, Reasons – 2, Slide -3) (08)

3. Cell biology - E – Electron Micrograph of organelles- Chloroplast, Mitochondria, Nucleus, Mitosis, Giant Chromosomes

– (Identification – 2, Diagram – 3, description – 3) (08)

4. Spotters – F, G, H, I, J and L(any seven of the following)

(Algae, Fungi, Bacteria, Virus, Bryophytes, Pteridophytes and Gymnosperms –permanent slides, book diagrams or wet preserved jar specimens, mentioned in the syllabus)

Cytology – photographs of cell organelles

Genetics – simple genetics problems

Plant biotechnology – tissue culture techniques : explants, callus, hardening

(Identification – 1, diagram – 1, Reasons – 2) (7 X 4) (28)

Total = 65

Record = 10

\_\_\_\_\_\_\_\_\_\_\_\_

Grand Total = 75

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| **SEMESTER: I**  **PART: III** | **23UZOOEP1**  **Zoology Practical– I** | **Credit: 1**  **Hours: 2** |

**Course Objectives**

|  |  |
| --- | --- |
| 1 | To identify the different groups of invertebrate animals by observing their external characteristics. |
| 2 | To understand the organs, organ system and their functions in lower animals. |
| 3 | To get knowledge about the different modes of life and their adaptation based on the environment. |
| 4 | Able to dissect and display the internal organs and mount the mouthparts and scales of invertebrates. |

**UNIT – I : Major Dissection :**

Cockroach: Circulatory system, Nervous system, Reproductive system. Leech : Nervous System, Reproductive system. Earthworm: Nervous System, Reproductive system. *Pila globosa*: Nervous system. Prawn: Nervous system (including Appendages).

**UNIT – II: Minor Dissection:**

Cockroach: Digestive system. Earthworm: Viscera, Lateral hearts.

*Pila globosa*: Digestive system (Including radula). Freshwater Mussel: Digestive system.

**UNIT – III: Mounting:**

Earthworm: Body setae; Pineal setae. *Pila globosa*: Radula. Freshwater muscle: Pedal ganglia.

**UNIT - IV: Mounting :**

Cockroach: Salivary apparatus, Mouth parts - Honey Bee, House fly and Mosquito mouth parts.

**UNIT - V: Spotters :(i).**

**Protozoa:** Amoeba, Paramoecium, Paramoecium Binary fission and Conjugation, Vorticella, Entamoeba histolytica, Plasmodium vivax **(ii). Porifera:** Sycon, Spongilla, Euspongia, Sycon - T.S & L.S, Spicules, Gemmule **(iii). Coelenterata:** Obelia – Colony & Medusa, Aurelia, Physalia, Velella, Corallium, Gorgonia, Pennatula **(iv). Platyhelminthes:** Planaria, Fasciola hepatica, Fasciola larval forms – Miracidium, Redia, Cercaria, Echinococcus granulosus, Taenia solium, Schistosoma haematobium **(v). Nemathelminthes:** Ascaris(Male & Female), Drancunculus, Ancylostoma, Wuchereria **(vi). Annelida:** Nereis, Aphrodite, Chaetopteurs, Hirudinaria, Trochophore larva **(vii). Arthropoda:** Cancer, Palaemon, Scorpion, Scolopendra, Sacculina, Limulus, Peripatus, Larvae - Nauplius, Mysis, Zoea, Mouth parts of male & female Anopheles and Culex, Mouthparts of Housefly and Butterfly. **(viii). Mollusca:** Chiton, Pila, Unio, Pteredo, Murex, Sepia, Loligo, Octopus, Nautilus, Glochidium larva **(ix). Echinodermata:** Asterias, Ophiothrix, Echinus, Clypeaster, Cucumaria, Antedon, Bipinnaria larva

**Expected Course Outcomes**

On completion of this course, students will;

|  |  |
| --- | --- |
| **1** | Identify and label the external features of different groups of invertebrate animals. |
| **2** | Illustrate and examine the circulatory system, nervous system and reproductive system of invertebrate animals. |
| **3** | Differentiate and compare the structure, function and mode of life of various groups of animals. |
| **4** | To compare and distinguish the dissected internal organs of lower animals. |
| **5** | Prepare and develop the mounting procedure of economically important invertebrates. |

**Text Books**

**(Latest Editions)**

1. Ekambaranatha Iyyar and T. N. Ananthakrishnan, 1995 A manual of Zoology Vol.I (Part 1, 2) S. Viswanathan, Chennai.
2. Ganguly, Sinha an d A dhikari , 2 0 11 . Biology of Animals: Volume I, New Central Book Agency; 3rd revised edition. 1008 pp.
3. Sinha, Chatterjee and Chattopadhyay, 2 0 1 4. Advanced Practical Zoology, Books & Allied Ltd; 3rd Revised edition, 1 07 0 pp.
4. Lal ,S. S, 2016 . Practical Zoology Invertebrate, Rastogi Publications.
5. Verma, P. S. 2010. A Manual of Practical Zoology: Invertebates, S Chand, 4 97pp.

**References Books**

1. Barnes, R.S.K., Calow, P., Olive, P.J.W., Golding, D.W. and Spicer, J.I. (2002). *The Invertebrates: A New Synthesis*, III Edition, Blackwell Science.
2. Barnes, R.D. (1982). *Invertebrate Zoology*, V Edition. Holt Saunders International Edition.
3. Barrington, E.J.W. (1979). *Invertebrate Structure and Functions*. II Edition, E.L.B.S. and Nelson
4. Boradale, L.A. and Potts, E.A. (1961). *Invertebrates: A Manual for the use of Students*. Asia Publishing Home.
5. Lal, S.S. 2005. A text Book of Practical Zoology: Invertebrate, Rastogi, Meerut

**Web Resources**

1. <https://nbb.gov.in/>
2. <http://www.agshoney.com/training.htm>
3. <https://icar.org.in/>
4. <http://www.csrtimys.res.in/>
5. <http://csb.gov.in/>
6. <https://iinrg.icar.gov.in/>
7. <https://www.nationalgeographic.com/animals/invertebrates/>

**Mapping with Programme Outcomes:**

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|  | **PO 1** | **PO 2** | **PO 3** | **PO 4** | **PO 5** | **PO 6** | **PO 7** | **PO 8** |
| **CO 1** | S |  |  | S | S | S | M |  |
| **CO 2** | M | S |  |  | M |  | L |  |
| **CO 3** |  |  | M | S |  | S |  |  |
| **CO 4** | S |  |  | S | S | M | S |  |
| **CO 5** |  |  | S |  |  | S |  | S |

**S-Strong(3) M-Medium (2) L-Low (1)**

|  |  |  |
| --- | --- | --- |
| **SEMESTER: I**  **PART: III** | **23UBIOEP1**  **Biochemistry Practical - I** | **Credit: 1**  **Hours: 2** |

**Learning objectives**

* Identify carbohydrates by qualitative test
* Estimate biomolecules volumetrically
* Estimate protein quantitatively

**I Qualitative analysis of carbohydrates**- 25Hrs

1. Monosaccharides-Glucose, Fructose
2. Disaccharides- Lactose, Maltose, Sucrose
3. Polysaccharides-Starch

**II Volumetric analysis 15 Hrs**

a) Estimation of ascorbic acid using 2,6dichlorophenolindophenol as link solution

b) Estimation of Glucose by Benedicts method

c)Estimation of Glycine by Sorenson Formal titration

**III Quantitative analysis**(Demonstration Expt)5 hrs

a)Colorimetric estimation of protein by Biuret method

**Course Outcome**

|  |  |  |
| --- | --- | --- |
| **CO** | **On completion of this course, students will be able to** | **Program**  **Outcomes** |
| CO1 | Qualitatively analyze and report the type of carbohydrate based on specific tests | PO1,PO2.PO3 |
| CO2 | Quantitatively estimate the carbohydrates, amino acids and ascorbic acid | PO1,PO2,PO3 |
| CO3 | Estimate protein by colorimetric method | PO1,PO2,PO3 |

**Text books**

1.Laboratory manual in Biochemistry, J. Jayaraman, 2nd edition, New Age International Publishers, 2011,

2. An Introduction to Practical Biochemistry, David T. Plummer, 3 rd edition, Tata McGraw-

Hill Publishing Company Limited, 2001.

3. Biochemical Methods, Sadasivam S and Manickam A, 4h edition, New Age International Publishers, 2016

**Mapping with Program Outcomes**

|  |  |  |  |  |  |  |  |  |  |  |
| --- | --- | --- | --- | --- | --- | --- | --- | --- | --- | --- |
|  | **PO 1** | **PO 2** | **PO 3** | **PO 4** | **PO 5** | **PO 6** | **PSO1** | **PSO2** | **PSO3** | **PSO4** |
| **CO 1** | 2 | 3 | 3 |  |  |  | 3 | 3 | 3 | 3 |
| **CO 2** | 2 | 3 | 3 |  |  |  | 3 | 3 | 3 | 3 |
| **CO 3** | 2 | 3 | 3 |  |  |  | 3 | 3 | 3 | 3 |

**S - Strong (3) M - Medi) L -Low(1)**

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| **SEMESTER: I**  **PART: IV** | **23UCHEF17**   **INTRODUCTORY CHEMISTRY** | **Credit: 2**  **Hours: 2** |

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| Objectives of the Course | To give insights into chemistry experiments for a beginner in   1. Lab safety and Nature of chemicals. 2. Types of titrations and Concentration terms. 3. Semi micro analysis and precipitation techniques. 4. Organic analysis 5. Gravimetric Principles |
| Course Outline | UNIT-I: LAB SAFETY, CHEMICALS AND GLASSWARE 1.1 laboratory hygiene and safety – first–aid techniques – general work culture inside the chemistry lab.  1.2 Nature of chemicals – toxic, corrosive, explosive, inflammable, carcinogenic, other hazardous chemicals – safe storing and handling of chemicals – disposal of chemical wastes.  1.3. Handling of glass wares- Calibration of pipette, standard measuring flask and burette. |
|  | UNIT-II: TITRIMETRIC METHODS OF ANALYSIS |
|  | 2.1 Definitions of Molarity and Normality. Primary and secondary standards, Criteria for primary standards-Preparation of standard solutions.  2.2 Concepts of Acids & Bases - pH of strong and weak acid solutions. Indicators-Theory and their choice..  2.3 Types of titrations- Acid-base Titrations, Redox Titrations, Precipitation Titrations and Complexometric Titrations- Principles and theory. |
|  | UNIT-III: SEMIMICRO METHODS 3.1. Identification of interfering & non-interfering acid radicals - removal of interfering radicals (any one test for each).  3.2 Separation of cations into groups-Reagents involved and their principle  3.3 Spot test analysis for ammonium, Pb, Cu, Mg, Mn and Ni. |
|  | UNIT-IV: BASICS OF ORGANIC ANALYSIS |
|  | 4.1 Preliminary and solubility tests for identifying organic compounds. Test for Aliphatic/Aromatic – Saturated/ Unsaturated compounds  4.2 Detection of Nitrogen, Sulphur and halogens  4.3– Test for functional groups: phenol, aldehyde, ketone, ester, carbohydrate, amine, amide & carboxylic acid (any one test for each). |
|  | UNIT V: GRAVIMETRIC METHODS  5.1 Gravimetric analysis- principle, theory and calculation.  5.2 Steps of a gravimetric analysis: precipitation, digestion, filtration, washing, drying and weighing.  5.2 Conditions for precipitation-choice of precipitants-advantages and disadvantages of using organic precipitants. |
| Extended  Professional  Component (is a  part of internal  component only,  Not to be included  in the external  examination  question paper) | Questions related to the above topics, from various competitive examinations UPSC/ JAM /TNPSC others to be solved (To be discussed during the Tutorial hours) |
| Skills acquired  from this course | Knowledge, Problem solving, Analytical ability, Professional Competency, Professional Communication and Transferable skills. |
| Recommended Text | 1. U.N. Dash, 2005, Analytical Chemistry: Theory and Practice, Sultan Chand and sons. Educational Publishers, 2nd Edition, New Delhi, 2. J.Bassett, R.C.Denney, G.H.Jerrey and J.Mendham, 1994,Vogel’s Text Book Of Inorganic Quantitative Analysis, ELBS, 5th Edition, London. 3. Gopalan R., Rangarajan K., Subramanian P.S. Elements of Analytical Chemistry, Sultan Chand & Sons, 2003 4. Svehla, 2012, Vogel’s Qualitative Analysis, Pearson Education, 7thEdition,New Delhi. 5. Venkateswaran V, Veeraswamy R, Kulandaivelu A R,1997,Basic Principles Of Practical Chemistry, Sultan Chand and Sons, 2nd Edition, New Delhi. 6. D.A. Skoog, D.M. West and F. J.Holler, 1990, Analytical chemistry,Saunders college publishing, 5th Edition, Philadelphia. |
| Reference Books | 1. Svehla, 2012, Vogel’s Qualitative Analysis, Pearson Education, 7thEdition,New Delhi. 2. Venkateswaran V, Veeraswamy R, Kulandaivelu A R,1997,Basic Principles Of Practical Chemistry, Sultan Chand and Sons, 2nd Edition, New Delhi |

|  |  |
| --- | --- |
|  | Knowledge, Problem solving, Analytical ability, Professional Competency, Professional Communication and Transferable skills. |
| Website and  e-learning source | 1. <https://www.tees.ac.uk/parttime_courses/engineering_&_construction/certificate_of_credit_foundation_process_chemistry_(by_flexible_open_learning).cfm> 2. <https://le.ac.uk/courses/chemistry-with-foundation-year-bsc/2023> 3. <https://www.researchgate.net/publication/345381808_Foundations_for_Teaching_Chemistry_Chemical_Knowledge_for_Teaching> 4. <https://yuli-elearning.com/mod/resource/view.php?id=738> 5. <https://pubs.acs.org/doi/10.1021/acs.jchemed.1c00666> |
| Course Learning Outcomes (for Mapping with POs and PSOs) On completion of the course the students should be able to  CO1: to understand laboratory safety and hygiene.  CO2: to understand principle of titrations.  CO3: to understand semi micro analysis.  CO4: to understand basics of organic compound analysis.  CO5: to understand about gravimetric analysis | |

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| --- | --- | --- | --- | --- | --- | --- | --- | --- | --- | --- |
|  | **PO1** | **PO2** | **PO3** | **PO4** | **PO5** | **PO6** | **PO7** | **PO8** | **PO9** | **PO10** |
| **CO1** | S | S | S | S | S | S | S | M | S | M |
| **CO2** | M | S | S | S | M | S | S | M | M | M |
| **CO3** | S | S | S | M | S | S | S | M | S | M |
| **CO4** | S | S | S | S | S | S | S | M | M | M |
| **CO5** | S | M | S | S | S | S | S | M | M | S |

|  |  |  |  |  |  |
| --- | --- | --- | --- | --- | --- |
| **CO/PSO** | **PSO1** | **PSO2** | **PSO3** | **PSO4** | **PSO5** |
| **CO1** | 2 | 2 | 2 | 2 | 2 |
| **CO2** | 2 | 2 | 2 | 2 | 2 |
| **CO3** | 2 | 2 | 2 | 2 | 2 |
| **CO4** | 2 | 2 | 2 | 2 | 2 |
| **CO5** | 2 | 2 | 2 | 2 | 2 |
| **Weightage** | 10 | 10 | 10 | 10 | 10 |
| **Weighted percentage of Course Contribution to Pos** | 2.0 | 2.0 | 2.0 | 2.0 | 2.0 |

Level of Correlation between PSO’s and CO’s

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| --- | --- | --- |
| **SEMESTER: II**  **PART: III**  **Core III** | **23UCHEC23**   **GENERAL CHEMISTRY-II** | **Credit: 4**  **Hours: 4** |

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| Objectives of the course | This course aims at providing an overall view of the   * chemistry of acids, bases and ionic equilibrium * properties of s and p-block elements * chemistry of hydrocarbons * applications of acids and bases * compounds of main block elements and hydrocarbons |
| Course Outline | UNIT-I  Acids, bases and Ionic equilibria  Concepts of Acids and Bases - Arrhenius concept, Bronsted-Lowry concept,  Lewis concept; Relative strengths of acids, bases and dissociation constant; dissociation of poly basic acids, ionic product of water, pH scale, pH of solutions; Degree of dissociation, common ion effect, factors affecting degree of dissociation; acid base indicators, theory of acid base indicators – action of phenolphthalein and methyl orange, titration curves - use of acid base indicators;  Buffer solutions – types, mechanism of buffer action in acid and basic buffer, Henderson-Hasselbalch equation;  Salt hydrolysis - salts of weak acids and strong bases, weak bases and strong acids, weak acids and weak bases - hydrolysis constant, degree of hydrolysis and relation between hydrolysis constant and degree of hydrolysis;  Solubility product - determination and applications; numerical problems involving the core concepts. |
|  | Unit-II  Chemistry of s - Block Elements  Hydrogen: Position of hydrogen in the periodic table. Alkali metals: Comparative study of the elements with respect to oxides, hydroxides, halides, carbonates and bicarbonates. Diagonal relationship of Li with Mg. Preparation, properties and uses of NaOH, Na2CO3, KBr, KClO3 alkaline earth metals. Anomalous behaviour of Be.  Chemistry of p- Block Elements (Group 13 & 14) preparation and structure of diborane and borazine. Chemistry of borax. Extraction of Al and its uses. Alloys of Al.  comparison of carbon with silicon. Carbon-di-sulphide – Preparation, properties, structure and uses. Percarbonates, per monocarbonates and per dicarbonates. |
|  | UNIT-III  Chemistry of p- Block Elements (Group 15-18)  General characteristics of elementsof Group 15; chemistry of H2N-NH2, NH2OH, HN3 and HNO3. Chemistry of PH3, PCl3, PCl5, POCl3, P2O5 and oxy acids of phosphorous (H3PO3 and H3PO4).    General properties of elements of group16 - Structure and allotropy of elements - chemistry of ozone - Classification and properties of oxides - oxides of sulphur and selenium – Oxy acids of sulphur (Caro’s and Marshall’s acids).    Chemistry of Halogens: General characteristics of halogen with reference to electro-negativity, electron affinity, oxidation states and oxidizing power. Peculiarities of fluorine. Halogen acids (HF, HCl, HBr and HI), oxides and oxy acids (HClO4). Inter-halogen compounds (ICl, ClF3, BrF5 and IF7), pseudo halogens [(CN)2 and (SCN)2] and basic nature of Iodine.    Noble gases: Position in the periodic table. Preparation, properties and structure of XeF2, XeF4, XeF6 and XeOF4; uses of noble gases - clathrate compounds. |
|  | UNIT-IV  Hydrocarbon Chemistry-I  Petroproducts: Fractional distillation of petroleum; cracking, isomerisation, alkylation, reforming and uses    Alkenes-Nomenclature, general methods of preparation – Mechanism of elimination reactions – E1 and E2 mechanism - factors influencing – stereochemistry – orientation – Hofmann and Saytzeff rules. Reactions of alkenes – addition reactions – mechanisms – Markownikoff’s rule, Kharasch effect, oxidation reactions – hydroxylation, oxidative degradation, epoxidation, ozonolysis; polymerization.    Alkadienes  Nomenclature - classification – isolated, conjugated and cumulated dienes; stability of conjugated dienes; mechanism of electrophilic addition to conjugated dienes - 1, 2 and 1, 4 additions; free radical addition to conjugated dienes– Diels–Alder reactions – polymerisation – polybutadiene, polyisoprene (natural rubber), vulcanisation, polychloroprene.    Alkynes  Nomenclature; general methods of preparation, properties and reactions; acidic nature of terminal alkynes and acetylene, polymerisation and isomerisation.    Cycloalkanes: Nomenclature, Relative stability of cycloalkanes, Bayer’s strain theory and its limitations. Conformational analysis of cyclohexane, mono and di substituted cyclohexanes. Geometrical isomerism in cyclohexanes. |

|  |  |
| --- | --- |
|  | UNIT-V  Hydrocarbon Chemistry - II  Benzene: Source, structure of benzene, stability of benzene ring, molecular orbital picture of benzene, aromaticity, Huckel’s (4n+2) rule and its applications. Electrophilic substitution reactions - General mechanism of aromatic electrophilic substitution - nitration, sulphonation, halogenation, Friedel-Craft’s alkylation and acylation. Mono substituted and disubstituted benzene - Effect of substituent – orientation and reactivity.  Polynuclear Aromatic hydrocarbons: Naphthalene – nomenclature, Haworth synthesis; physical properties, reactions – electrophilic substitution reaction, nitration, sulphonation, halogenation, Friedel – Crafts acylation & alkylation, preferential substitution at  - position – reduction, oxidation – uses.  Anthracene – synthesis by Elbs reaction, Diels – Alder reaction and Haworth synthesis; physical properties; reactions - Diels-Alder reaction, preferential substitution at C-9 and C-10; uses. |
| Extended  Professional Component (is a part of internal component only, Not to be included in the external examination question paper) | Questions related to the above topics, from various competitive examinations  UPSC/JAM /TNPSC others to be solved  (To be discussed during the Tutorial hours) |
| Skills acquired from this course | Knowledge, Problem solving, Analytical ability, Professional Competency, Professional Communication and Transferable skills. |
| Recommended  Text | 1. Madan R D, Sathya Prakash, (2003), Modern Inorganic Chemistry, 2nded,   S.Chand and Company, New Delhi.   1. Sathya Prakash, Tuli G D,Basu S K and Madan R D, (2003), Advanced Inorganic Chemistry, 17th ed., S.Chand and Company, New Delhi. 2. Bahl B S, Arul Bhal, (2003), Advanced Organic Chemistry, 3rd ed., S.Chand and Company, New Delhi. 3. Tewari K S, Mehrothra S N and Vishnoi N K, (1998), Text book of Organic Chemistry, 2nd ed., Vikas Publishing House, New Delhi. 4. Puri B R, Sharma L R, (2002), Principles of Physical Chemistry, 38th ed., Vishal Publishing Company, Jalandhar. |
| Reference  Books | 1. Maron S H and Prutton C P, (1972), Principles of Physical Chemistry, 4th ed., The Macmillan Company, Newyork. 2. Barrow G M, (1992), Physical Chemistry, 5th ed., Tata McGraw Hill, New Delhi. 3. Lee J D, (1991), Concise Inorganic Chemistry, 4thed., ELBS William Heinemann, London. 4. Huheey J E, (1993), Inorganic Chemistry: Principles of Structure and Reactivity, 4th ed., Addison Wesley Publishing Company, India. 5. Gurudeep Raj, (2001), Advanced Inorganic Chemistry Vol – I, 26th ed., Goel Publishing House, Meerut. 6. Agarwal O P, (1995), Reactions and Reagents in Organic Chemistry, 8thed., Goel Publishing House,Meerut. |
| Website and e-learning source | https://onlinecourses.nptel.ac.inhttp://cactus.dixie.edu/smblack/chem1010/lec  ture\_notes/4B.html http://www.auburn.edu/~deruija/pdareson.pdfhttps://swayam.gov.in/course/64 -atomic-structure-and-chemical-bonding  MOOC components  http://nptel.ac.in/courses/104101090/  Lecture 1: Classification of elements and periodic properties http://nptel.ac.in/courses/104101090/ |

Course Learning Outcomes (for Mapping with POs and PSOs)

On completion of the course the students should be able to

CO1: explain the concept of acids, bases and ionic equilibria; periodic properties of s and p-block elements, preparation and properties of aliphatic and aromatic hydrocarbons CO2: discuss the periodic properties of sand p- block elements, reactions of aliphatic and aromatic hydrocarbons and strength of acids

CO3: classify hydrocarbons, types of reactions, acids and bases, examine the properties s and p-block elements, reaction mechanisms of aliphatic and aromatic hydrocarbons CO4: explain theories of acids, bases and indicators, buffer action and important compounds of s-block elements

CO5: assess the application of hard and soft acids indicators, buffers, compounds of s and p-block elements and hydrocarbons

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| --- | --- | --- | --- | --- | --- | --- | --- | --- | --- | --- |
|  | PO1 | PO2 | PO3 | PO4 | PO5 | PO6 | PO7 | PO8 | PO9 | PO10 |
| CO1 | S | S | S | S | S | S | S | M | S | M |
| CO2 | M | S | S | S | M | S | S | M | M | M |
| CO3 | S | S | S | M | S | S | S | M | S | M |
| CO4 | S | S | S | S | S | S | S | M | M | M |
| CO5 | S | M | S | S | S | S | S | M | M | S |

CO-PO Mapping (Course Articulation Matrix)

|  |  |  |  |  |  |
| --- | --- | --- | --- | --- | --- |
| CO /PO | PSO1 | PSO2 | PSO3 | PSO4 | PSO5 |
| CO1 | 3 | 3 | 3 | 3 | 3 |
| CO2 | 3 | 3 | 3 | 3 | 3 |
| CO3 | 3 | 3 | 3 | 3 | 3 |
| CO4 | 3 | 3 | 3 | 3 | 3 |
| CO5 | 3 | 3 | 3 | 3 | 3 |
| Weightage | 15 | 15 | 15 | 15 | 15 |
| Weighted percentage of  Course Contribution to Pos | 3.0 | 3.0 | 3.0 | 3.0 | 3.0 |

Level of Correlation between PSO&CO

|  |  |  |
| --- | --- | --- |
| **SEMESTER: II**  **PART: II**  **Core IV** | **23UCHEP24**  **QUALITATIVE ORGANIC ANALYSIS AND PREPARATION OF ORGANIC COMPOUNDS** | **Credit: 5**  **Hours: 4** |

|  |  |
| --- | --- |
| Objectives of the course | This course aims at providing knowledge on   * laboratory safety * handling glass wares * analysis of organic compounds * preparation of organic compounds |
| Course Outline | UNIT I  Safety rules, symbols and first-aid in chemistry laboratory  Basic ideas about Bunsen burner, its operation and parts of the flame.  Chemistry laboratory glassware –basis information and uses |
| Unit II  Qualitative Organic Analysis  Preliminary examination, detection of special elements - nitrogen, sulphur and halogens  Aromatic and aliphatic nature, Test for saturation and unsaturation, identification of functional groups using solubility tests  Confirmation of functional groups   * monocarboxylic acid, dicarboxylic acid * monohydric phenol, polyhydric phenol * aldehyde, ketone, ester * carbohydrate (reducing and non-reducing sugars) * primary, secondary, tertiary amine * monoamide, diamide, thioamide * anilide, nitro compound * Preparation of derivatives for functional groups |
| UNIT III  Preparation of Organic Compounds   1. Nitration - picric acid from Phenol 2. Halogenation - p-bromo acetanilide from acetanilide 3. Oxidation - benzoic acid from Benzaldehyde 4. Microwave assisted reactions in water: 5. Methyl benzoate to Benzoic acid 6. Salicylic acid from Methyl Salicylate 7. Rearrangement - Benzil to Benzilic Acid 8. Hydrolysis of benzamide to Benzoic Acid |
|  | Unit-IV  Separation and Purification Techniques (Not for Examination)   1. Purificationof organic compounds by crystallization (from water / alcohol) and distillation 2. Determination of melting and boiling points of organic compounds.   3.Steam distillation - Extraction of essential oil from citrus fruits/eucalyptus leaves.  4. Chromatography (any one) (Group experiment)  (i) Separation of amino acids by Paper Chromatography  (ii)Thin Layer Chromatography - mixture of sugars / plant pigments  /permanganate dichromate.  (iii) Column Chromatography - extraction of carotene, chlorophyll and xanthophyll from leaves / separation of anthracene - anthracene picrate.   1. Electrophoresis – Separation of amino acids and proteins.   (Demonstration)   1. Isolation of casein from milk/Determination of saponification value of oil or fat/Estimation of acetic acid from commercial vinegar. (Any one Group experiment) (4,5& 6–not for ESE) |
| Reference  Books | 1. Venkateswaran, V.; Veeraswamy, R.; Kulandaivelu, A.R. Basic Principles of Practical Chemistry, 2nd ed.; Sultan Chand: New Delhi, 2012. 2. Manna, A.K. Practical Organic Chemistry, Books and Allied: India, 2018. 3. Gurtu, J. N; Kapoor, R. Advanced Experimental Chemistry (Organic), Sultan Chand: New Delhi, 1987. 4. Furniss,B. S.; Hannaford, A. J.; Smith, P. W. G.; Tatchell, A.R. Vogel’s Textbook of Practical Organic Chemistry, 5th ed.; Pearson: India,1989. |
| Website and e-learning source | <https://www.vlab.co.in/broad-area-chemical-sciences> |
| Scheme of Valuation Max. marks(75)  Record :10 Marks  Preparation :15 Marks  Recrystallization :05 Marks  Organic Qualitative Analysis :45 Marks  Preliminary Test :05 Marks  Detection of Elements :05 Marks  Detection of Functional Group :05 Marks  Identification of the compound :05 Marks  Confirmatory Test :15 Marks  Derivatives preparation and its m.pt determination :10 Marks | |

Course Learning Outcomes (for Mapping with POs and PSOs)

On completion of the course the students should be able to

CO1: observe the physical state, odour, colour and solubility of the given organic compound.

CO2: identify the presence of special elements and functional group in an unknown organic compound performing a systematic analysis.

CO3: compare mono and dicarboxylic acids, primary, secondary and tertiary amines, mono and diamides, mono and polyhydric phenols, aldehyde and ketone, reducing and non- reducing sugars and explain the reactions behind it.

CO4: exhibit a solid derivative with respect to the identified functional group.





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| --- | --- | --- | --- | --- | --- |
| CO /PO | PSO1 | PSO2 | PSO3 | PSO4 | PSO5 |
| CO1 | 3 | 3 | 3 | 3 | 3 |
| CO2 | 3 | 3 | 3 | 3 | 3 |
| CO3 | 3 | 3 | 3 | 3 | 3 |
| CO4 | 3 | 3 | 3 | 3 | 3 |
| Weightage | 12 | 12 | 12 | 12 | 12 |
| Weighted percentage of  Course Contribution to  Pos | 3.0 | 3.0 | 3.0 | 3.0 | 3.0 |

Level of Correlation between PSO’s and CO’s

|  |  |  |
| --- | --- | --- |
| **SEMESTER: II**  **PART: II**  **Core IV** | **23UMATE25**  **MATHEMATICS – II** | **Credit: 3**  **Hours: 5** |

**UNIT-I:TRIGONOMETRY**

Expansions of sin n θ, cos n θ, sinnθ,cosnθ, tannθ – Expansions of sinθ, cosθ, tanθ in terms of θ

Unit-I: Chap: 6 (6.1,6.1.1 to 6.1.3)

**UNIT-II: PARTIAL DIFFERENTIAL EQUATIONS**

Formation of partial differential equations, elimentary partial differential equations- Lagranges equations.

Unit-II: Chap:6 (6.1,6.1.1, 6.4).

**UNIT-III: VECTOR DIFFRENTIATION**

Vector functions- Scalar and vector point functions- Directional derivatives –Unit vector normal to a surface – angle between the surfaces-divergence, Gradient of a scalar point function- Divergence and curl of a vector point function.

Unit-III Section 8.1,8.2,8.3, 8.4).

**UNIT-IV: VECTOR INTEGRATION**

Green’s theorem in the plane-Gauss divergence theorem- [without proofs] ,Stoke’s theorem

( Statement only)

Unit-IV:Section(8.6.1, to 8.6.3).

**UNIT-V: FINITE DIFFERENCES**

Operator E, Relation between  and E – Interpolation – Newton – Gregory forward & backward formulae for interpolation-Lagrange’s interpolation formula for unequal intervals(without proof) .

Unit-V:Sec(5.1,5.2).

**TEXT BOOK:**

1.P. Duraipandian and S. Udayabaskaran(1997), “Allied Mathematics”, Vol I & II. Chennai:

Muhil Publishers.

Unit-I: Chap: 6 (6.1,6.1,1 to 6.1.3), Vol I,

Unit-II: Chap:6 (6.1,6.1.1,6.4), Vol II,

Unit-IIISec(8.1,8.2,8.3,8.4),Vol I,

Unit-IV:Sec(8.6.1, - 8.6.3), Vol I,

Unit-V:Sec(5.1,5.2), Vol II.

**REFERENCE BOOKS:**

1. P. Balasubramanian and K. G. Subramanian. 1997, “Ancillary Mathematics”, Vol I & II. New Delhi: Tata McGraw Hill.
2. S.P.Rajagopalan and R.Sattanathan(2005), “Allied Mathematics”, Vol I & II. New Delhi: Vikas Publications.
3. P. R. Vittal (2003), “Allied Mathematics”,Chennai: Marghan Publications.
4. P.Kandhasamy, K. Thilagavathy (2003), “Allied Mathematics” Vol I & II, New Delhi: Tata McGraw Hill.

**Course Outcomes:**

On successful completion of the course, the students will be able to

**CO1**: Attain knowledge on finding the expansions of trigonometric functions and concept of hyperbolic and inverse hyperbolic functions.

**CO2**: Provide a basic knowledge of Partial Differential equations and develops knowledge on handle practical problems.

**CO3**: Adopt techniques in solving problems involving vector and scalar functions

**CO4**: Provide skills on finding derivatives and gradients on vector differentiation and

Integration.

**CO5**: Understand the applications of differentiation and integration in real life situation.

**Outcome Mapping:**

|  |  |  |  |  |  |
| --- | --- | --- | --- | --- | --- |
| CO / PO | PO1 | PO2 | PO3 | PO4 | PO5 |
| CO1 | 3 | 3 | 3 | 2 | 2 |
| CO2 | 3 | 2 | 3 | 3 | 2 |
| CO3 | 3 | 3 | 3 | 3 | 2 |
| CO4 | 3 | 3 | 3 | 3 | 2 |
| CO5 | 2 | 3 | 3 | 3 | 2 |

**1-Low 2-Moderate 3- High**

|  |  |  |
| --- | --- | --- |
| **SEMESTER: II**  **PART: III** | **23UBOTE25**  **BOTANY-II**  **(Elective)** | **Credit: 3**  **Hours: 4** |

**Learning Objective (LO):**

|  |  |
| --- | --- |
| **LO1** | To be familiar with the basic concepts and principles of plant systematics. |
| **LO2** | Learn the importance of plant anatomy in plant production systems. |
| **LO3** | Understand the mechanism underling the shift from vegetative to reproductive phase. |
| **LO4** | To learn about the physiological processes that underlie plant metabolism. |
| **LO5** | To know the energy production and its utilization in plants. |

**Unit – 1: MORPHOLOGY OF FLOWERING PLANTS**

Plant and its parts. Structure and function of root and stem. Leaf and its parts. Leaf types- simple and compound. Phyllotaxy and types. Inflorescence - Racemose, Cymose and Special types. Terminology with reference to flower description.

**Unit – 2 : TAXONOMY**

Study of the range of characters and plants of economic importance in the following families: Rutaceae, Caesalpiniaceae, Asclepiadaceae, Euphorbiaceae and Cannaceae

**Unit – 3: ANATOMY**

Tissue and tissue systems: Simple and complex tissues. Anatomy of monocot and dicot roots - anatomy of monocot and dicot stems - anatomy of dicot and monocot leaves.

**Unit – 4: EMBRYOLOGY**

Structure of mature anther and ovule - Types of ovules, structure of embryo sac, pollination -double fertilization, structure of dicotyledonous and monocotyledonous seeds.

**Unit – 5: PLANT PHYSIOLOGY**

Absorption of water, photosynthesis - light reaction - Calvin cycle; respiration - Glycolysis - Krebs cycle - electron transport system. Growth hormones - auxins and cytokinins and their applications.

**Course Outcomes (CO)**

At the end of the course, the student will be able to

|  |  |
| --- | --- |
| **CO1** | Understand the fundamental concepts of plant anatomy and  embryology. |
| **CO2** | Analyze and recognize the different organs of plants and secondary  growth. |
| **CO3** | Understand water relation of plants with respect to various  physiological processes |
| **CO4** | Classify aerobic and anaerobic respiration. |
| **CO5** | Classify plant systematics and recognize the importance of  herbarium and virtual herbarium. |

**Recommended Texts**

1. Sharma, O.P. 2017. Plant Taxonomy. (II Edition).The McGraw Hill Companies.
2. Bhojwani, S.S. Bhatnagar, S.P and Dantu, P.K. 2015. The Embryology of Angiosperms

(6th revised and enlarged edition). Vikas Publishing House, New Delhi.

1. Maheshwari, P. 1963. Recent Advances in Embryology of Angiosperms. Intl. Soc.

Plant Morphologists, New Delhi.

1. Salisbury, F. B.C.W. Ross.1991. Plant Physiology. Wassworth Pub. Co. Belmont.
2. Ting, I.P. 1982. Plant Physiology. Addison Wesley Pb. Philippines.

**Reference books**

1. Lawrence.G.H.M. 1985. An Introduction to Plant Taxonomy, Central Book Depot, Allahabad.
2. Bhojwani, S.S and Bhatnagar, S.P. 2000. The Embryology of Angiosperms (4th revised and

enlarged edition). Vikas Publishing House, New Delhi.

1. Pandey, B.P. 2012. Plant Anatomy. S Chand Publishing.
2. Jain, VK. 2006. Fundamentals of Plant Physiology, S. Chand and Company Ltd.
3. [Rajni Gupta](https://www.amazon.in/s/ref=dp_byline_sr_book_1?ie=UTF8&field-author=Rajni+Gupta&search-alias=stripbooks). 2012. Plant Taxonomy: Past, Present and Future.  [Vedams (P) Ltd. New Delhi.](https://www.abebooks.com/vedams-ebooks-p-ltd-new-delhi/573945/sf" \o "Vedams eBooks (P) Ltd)
4. Jain, V.K. 2006. Fundamentals of Plant Physiology, S.Chand and Company Ltd., New Delhi.
5. Verma, S.K. 2006. A Textbook of Plant Physiology, S.K.Chand & Co., New Delhi.

**Web Resources**

1. <https://books.google.co.in/books/about/Plant_Taxonomy.html?id=0bYs8F0Mb9gC&redir_esc=y>
2. <https://books.google.co.in/books/about/PLANT_TAXONOMY_2E.html?id=Roi0lwSXFnUC&redir_esc=y>

##### [https://archive.org/EXPERIMENTS/plantanatomy031773mbp](https://archive.org/details/plantanatomy031773mbp)

1. <https://www.amazon.in/Embryology-Angiosperms-6th-S-P-Bhatnagar-ebook/dp/B00UN5KPQG>
2. <https://www.crcpress.com/Plant-Physiology/Stewart-Globig/p/book/9781926692692>

**Mapping with Programme Outcomes:**

|  |  |  |  |  |  |  |  |  |  |  |
| --- | --- | --- | --- | --- | --- | --- | --- | --- | --- | --- |
| **COs** | **PO1** | **PO2** | **PO3** | **PO4** | **PO5** | **PSO1** | **PSO2** | **PSO3** | **PSO4** | **PSO5** |
| **CO 1** | S | S | S | S | S | S | S | S | S | S |
| **CO 2** | S | S | S | S | S | S | S | S | S | S |
| **CO 3** | M | S | S | S | S | L | S | S | S | S |
| **CO 4** | S | S | M | S | S | S | S | M | S | M |
| **CO 5** | S | M | M | M | M | M | M | L | M | M |

**S – Strong; M – Medium; L – Low**

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| --- | --- | --- |
| **SEMESTER: II**  **PART: III** | **23UZOOE25**  **Zoology – II** | **Credit: 2**  **Hours: 3** |

|  |  |
| --- | --- |
| 1 | To enable students to learn basic concepts relating to aspects of respiratory, circulatory, excretory, nervous and sensory physiology. |
| 2 | To enable students to comprehend the processes involved during development |
| 3 | To enable students to learn basic concepts of immunity and the working of immune organs and familiarize them with the recommended vaccination schedule |
| 4 | To enable students to comprehend the basic concepts of human genetics and patterns of inheritance |
| 5 | To enable students to learn about aspects of animal behaviour such as foraging, courtship, nest construction, parental care and learning |

Unit – I: Respiration- Respiratory pigments and transport of gases. Mechanism of blood clotting. Types of excretory products – Ornithine cycle. Structure of neuron –Conduction of nerve impulse, Mechanism of vision and hearing.

Unit – II: Fertilization, Cleavage, Gastrulation and Organogenesis of Frog; Placentation in mammals

Unit – III: Innate and Acquired - Active and Passive; Antigens and Antibodies; Immunological organs–responses in humans; Vaccination schedule

Unit – IV: Human Genetics: Human Chromosomes – Sex Determination in Humans; Patterns of Inheritance: Autosomal Dominant, Autosomal Recessive, X-linked , Y-linked, Mitochondrial, Multiple Allelic and Polygenic; Genetic Counseling

Unit - V: Animal Behaviour: Foraging, Courtship Behaviour, Shelter and Nest Construction, Parental Care, Learning Behaviour

**Expected Course Outcomes**

On completion of this course, students will be able to:

|  |  |
| --- | --- |
| **1** | Recall the parts and working of body organs and developmental stages, name the patterns of inheritance and list different types of animal behaviour |
| **2** | Analyse the different developmental stages |
| **3** | Analyse the working of body and immune systems |
| **4** | Analyse the different patterns of inheritance |
| **5** | Relate the behaviour of animals to physiology. Analyse the different types of behaviour |

**Text Books (Latest Editions)**

* + - 1. Verma P.S. & Agarwal - Developmental Biology, Chordata embryology S. Chand & Co.

**References Books**

**(Latest editions, and the style as given below must be strictly adhered to)**

Owen, J. A., Punt, J. & Stranford, S. A. Kuby Immunology. New York: W.H. Freeman & Company.

Klug, W. S., Cummings, M. R. & Spencer, C - Concepts of Genetics. (12th ed.). New Jersey: Pearson Education.

Mathur, R. Animal Behaviour. Meerut: Rastogi.

Verma P.S. & Agarwal Developmental Biology, Chordata embryology. S.Chand & Co.

**Outcome Maping**

|  |  |  |  |  |  |  |  |  |
| --- | --- | --- | --- | --- | --- | --- | --- | --- |
|  | **PO 1** | **PO 2** | **PO 3** | **PO 4** | **PO 5** | **PO 6** | **PO 7** | **PO 8** |
| **CO 1** | S |  | S |  | M |  | S | S |
| **CO 2** | M | S |  |  |  |  |  |  |
| **CO 3** |  | S | M | S |  | S | M |  |
| **CO 4** | S |  |  | S | S | M |  |  |
| **CO 5** |  |  | S |  |  |  |  | S |

**S-Strong M-Medium L-Low**

|  |  |  |
| --- | --- | --- |
| **SEMESTER: II**  **PART: III** | **23UBIOE25**  **Biochemistry- II** | **Credit: 2**  **Hours: 3** |

**Learning objectives**

The objectives of this course are to

* Impart knowledge on the classification, properties and characterization of lipids.
* Comprehend the metabolism of Lipids
* Acquaint with the structure, properties and functions of nucleic acids
* Learn about the enzyme kinetics and inhibition
* Study the importance of Hormones

**Module I :**Lipids–Bloor’s classification of lipids- simple lipids, fatty acids (saturated and unsaturated), compound lipids, derived lipids.Properties of lipids- reduction, oxidation, halogenation,saponification and rancidity .Classification andfunctions of phospholipids, Cholesterol – structure and biological importance.12 Hrs

**Module II** :Lipid metabolism- Oxidation of fatty acids(Palmitic acid ) – Beta oxidation-Role of carnitine,energetics , alpha oxidation and omega oxidation.Biosynthesis of saturated fatty acids.12 Hrs

**Module III :**Purine and pyrimidine bases, nucleosides, nucleotides, polynucleotides, DNA structure, various types, properties- absorbance, effect of temperature. Different types of RNA, structure and function, Genetic code. 12 Hrs

**Module III :**Enzymes - Nomenclature, IUB system of enzyme classification,active site, specificity, isoenzymes, units of enzyme activity factors affecting enzyme activity- substrate concentration, pH, temperature.Enzyme Kinetics- Michaelis and Menten equation.Lineweaver- Burk plot. Enzyme inhibition, competitive, uncompetitive and andnon competitive inhibition 12Hrs

**Module V:** Hormones -classification,Biological functions of Insulin, Thyroid and Reproductive hormones . 12Hr

**Course Outcome**

|  |  |  |
| --- | --- | --- |
| **CO** | **On completion of this course, students will be able to** | **Program**  **Outcomes** |
| CO1 | Elaborate on classification, structure, properties, functions and characterization of lipids | PO1 |
| CO2 | Discuss the metabolism of lipids and its importance | PO1 |
| CO3 | Explain about structure, properties and functions of nucleic acids | PO1 |
| CO4 | Derive Michaelis Menten equation and concepts of enzyme inhibition | PO1,PO3 |
| CO5 | Classify the Hormones and its biological functions | PO1,PO4 |

**Text books**

1.Satyanarayan,U (2014) Biochemistry (4th ed), Arunabha Sen Books & Allied (P) Ltd, Kolkata.

2.Jain J.L.(2007) Fundamentals of Biochemistry,S.Chand publishers

**Reference books**

**1.** David L.Nelson and Michael M.Cox (2012) Lehninger Principles of Biochemistry (6th ed) W.H. Freeman.

**2.** Voet.D & Voet. J.G (2010) Biochemistry , (4th ed), John Wiley & Sons, Inc.

**3.** Lubert Stryer (2010) Biochemistry,(7th ed), W.H.Freeman

**Web sources**

1.onlinecourses.swayam2.ac.in/cec20\_bt12

2 onlinecourses.swayam2.ac.in/cec20\_bt19

**Mapping with Program Outcomes**

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| --- | --- | --- | --- | --- | --- | --- | --- | --- | --- | --- |
|  | **PO 1** | **PO 2** | **PO 3** | **PO 4** | **PO 5** | **PO 6** | **PSO1** | **PSO2** | **PSO3** | **PSO4** |
| **CO 1** | 3 |  |  |  |  |  | 3 |  |  | 3 |
| **CO 2** | 3 |  |  |  |  |  | 3 |  |  | 3 |
| **CO 3** | 3 |  | 3 |  |  |  | 3 |  |  | 3 |
| **CO 4** | 3 |  |  | 3 |  |  | 3 |  |  | 3 |
| **CO5** | 3 |  |  |  |  |  | 3 | 3 |  | 3 |

**S - Strong (3) M - Medium (2) L -Low(1)**

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| --- | --- | --- |
| SEMESTER –II  PART – III | Elective  23UBOTEP2: Botany Practical II | CREDITS: 1  HOURS: 30 |

**Course Objectives**

1. To enhance information on the identification of taxonomical plant
2. To be familiar with the basic concepts and principles of plant systematics.
3. Understanding of reproduction and development of angiosperms
4. To understand the internal organization of Angiopserms
5. To learn about the physiological processes that underlie plant metabolism.

**EXPERIMENTS**

1. To identify Angiosperm root, stem, leaf, flowers and fruits based on morphology
2. To describe in technical terms, plants belonging to any of the family prescribes and to identify the family.
3. To dissect a flower, construct floral diagram and write floral formula.
4. Demonstration experiments
   1. Ganong’s Light screen
   2. Ganong’s respiroscope
5. To make suitable micro preparations of anatomy materials prescribed in the syllabus.
6. Spotters - Angiosperm morphology, anatomy, Embryology and Physiology

**Bonafide record of practical work done should be submitted for the practical examination**

**Course outcomes:**

On completion of this course, the students will be able to:

1. Understand external structure of angiosperms
2. To study the classical taxonomy with reference to different parameters.
3. Understand the fundamental concepts of plant anatomy and embryology
4. To study the effect of various physical factors on photosynthesis.
5. Understand simple experiments in plant Physiology

**Recommended texts**

1. Sharma, O.P. 2017. Bryophyta, MacMillan India Ltd, New Delhi.

2. Sharma, O.P. 2012. Pteridophyta, Tata McGraw-Hills Ltd, New Delhi.

3. Subramaniam, N.S. 1996. Laboratory Manual of Plant Taxonomy. Vikas Publishing House Pvt. Ltd., New Delhi.

4. Benjamin, A. Pierce. 2012. Genetics- A conceptual Approach. W.H. Freeman and Company, New York, England.

5.Noggle G.R and G.J. Fritz. 2002. Introductory Plant Physiology. Prentice Hall of India, New Delhi.

**Reference books**

1. Strickberger, M.W. 2005. Genetics (III Ed). Prentice Hall, New Delhi, India.
2. Nancy Serediak and M. Huynh. 2011. Algae identification lab Guide. Accompanying manual to algae identification field guide, Ottawa Agriculture and Agri food Canada publisher.
3. Mohammed Gufran Khan, Shite Gatew and Bedilu Bekele. 2012. Practical manual for Bryophytes and Pteridophytes. Lambert Academic Publishing.
4. Aler Gingauz. 2001. Medicinal Chemistry. Oxford University Press & Wiley Publications.
5. Steward, F.C. 2012. Plant Physiology Academic Press, US

**Web Resources**

1. https://www.amazon.in/Practical-Manual-Pteridophyta-Rajan-Sundara/dp/8126106883
2. [https://www.google.co.in/books/edition/Gymnosperms/3YrT5E3Erm8C?hl=en&gbpv=1&dq=gy](https://www.google.co.in/books/edition/Gymnosperms/3YrT5E3Erm8C?hl=en&gbpv=1&dq=gymnosperms&printsec=frontcover) [mnosperms&printsec=frontcover](https://www.google.co.in/books/edition/Gymnosperms/3YrT5E3Erm8C?hl=en&gbpv=1&dq=gymnosperms&printsec=frontcover)
3. <https://www.amazon.in/Manual-Practical-Bryophyta-Suresh-Kumar/dp/B0072GNFX4>

OUTCOME MAPPING

|  |  |  |  |  |  |
| --- | --- | --- | --- | --- | --- |
|  | **PO1** | **PO2** | **PO3** | **PO4** | **PO5** |
| **CO1** | 2 | 3 | 1 | 3 | 2 |
| **CO2** | 2 | 3 | 2 | 2 | 3 |
| **CO3** | 2 | 1 | 3 | 2 | 3 |
| **CO4** | 1 | 3 | 3 | 2 | 2 |
| **CO5** | 2 | 2 | 3 | 1 | 3 |

**BOTANY PRACTICAL II**

Time : 3 Hours Max. Marks : 75

**PRACTICAL QUESTION PAPER**

1. Identify the given specimens –A to its respective family, draw MLS of the flower and describe it in technical terms.

(Identification of family – 2, MLS diagram – 3, technical description – 4) (09)

2. Identify the given specimen –B, to its respective family, construct the floral diagram and write the floral formula.

(Identification of family – 2, floral diagram – 3, floral formula – 2) (07)

3. Make suitable micro preparations of the given specimens C.

Submit the slides for valuation. Identify the specimens, draw diagrams and give reasons. (Identification – 1, diagram – 2, Reasons – 2, Slide -2) (07)

4. Comment on the Physiology setup – D Write the aim, materials required , Procedure, Results and Inference

( Aim-1, Materials required -1, Procedure -2 , Results and Inference -3) (07)

4. Spotters – E, F, G, H, I, J, K and L.

(Identification – 1, diagram – 2, Reasons – 2) (7 X 5) (35) \_\_\_\_\_\_\_\_\_\_\_\_

Total = 65

Record = 10

\_\_\_\_\_\_\_\_\_\_\_\_

Grand Total = 75

\_\_\_\_\_\_\_\_\_\_\_\_

BOTANY PRACTICAL II

KEY & SCHEME OF VALUATION

1. Taxonomy - A – MLS of the flower (from any one family mentioned in the syllabus)

(Identification of family – 2, MLS diagram – 3, technical description – 4) (09)

2. Taxonomy - B – Floral diagram and floral formula (from any one family mentioned in the syllabus) (Identification of family – 2, floral diagram – 3, floral formula – 2) (07)

3.. Anatomy - C : Dicot and monocot – stem, root and leaf.

(Identification – 1, diagram – 2, Reasons – 2, Slide -2) (07)

4. Physiology Set up D - Osmosis – thistle funnel experiment, Photosynthesis – Beaker and Funnel experiment, Ganong’s light screen and Ganong’s respire scope

( Aim-1, Materials required -1, Procedure -2 , Results and Inference -3) (07)

5. Spotters –E, F, G, H, I, J , and K (any seven of the following) (08)

Morphology – vegetative and reproductive morphological parts

Anatomy – simple and complex tissues, dicot, monocot root and leaf

Embryology – ovules, anther T.S.

Physiology - Osmosis – thistle funnel experiment, Photosynthesis – Beaker and Funnel experiment, Ganong’s light screen and Ganong’s respire scope experimental setup.

(Identification – 1, diagram – 2, Reasons – 2) (7 X 5) (35)

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Total = 65

Record = 10

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Grand Total = 75

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| **SEMESTER: II**  **PART: III** | **23UBIOEP2**  **Biochemistry Practical - II** | **Credit: 2**  **Hours: 3** |

**Learning objectives**

The objectives of this course are to

* Identify amino acids by qualitative test
* Prepare biomolecules from its sources
* Estimate phosphorus quantitatively

I.**Qualitative analysis of amino acids**

a) Arginine b)Cysteine c) Tryptophan d)Tyrosine e) Histidine

II. **Biochemical preparations**

a) Preparation of casein from milk.

b)Preparation of starch from potato.

c)Preparation of albumin from egg.

**IIIGroup Experiment**

Determination of Iodine/ Saponification number of an edible oil(Demonstration) .

**Course Outcome**

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| --- | --- | --- |
| **CO** | **On completion of this course, students will be able to** | **Programme**  **Outcome** |
| CO1 | Qualitatively analyze the amino acids and report the type of amino acids based on specific tests | PO1,PO2,PO3 |
| CO2 | Prepare the macronutrients from the rich sources. | PO1,PO2,PO3 |
| CO3 | Check the quality of edible oil | PO1,PO2,PO3 |

**Text books**

1.Laboratory manual in Biochemistry, J. Jayaraman, 2nd edition, NewAge International Publishers, 2011,

2. An Introduction to Practical Biochemistry, David T. Plummer, 3 rd edition, Tata McGraw-Hill Publishing Company Limited, 2001.

**Reference books**

1. Biochemical Methods, Sadasivam S and Manickam A, 4h edition, NewAge International Publishers, 2016

2. Essentials of Food and Nutrition, Vol. I &amp; II, M.S. Swaminathan.

**Mapping with Program Outcomes**

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| --- | --- | --- | --- | --- | --- | --- | --- | --- | --- | --- |
|  | **PO 1** | **PO 2** | **PO 3** | **PO 4** | **PO 5** | **PO 6** | **PSO1** | **PSO2** | **PSO3** | **PSO4** |
| **CO 1** | 2 | 3 | 3 |  |  |  | 3 | 3 | 3 | 3 |
| **CO 2** | 2 | 3 | 3 |  |  |  | 3 | 3 | 3 | 3 |
| **CO 3** | 2 | 3 | 3 |  |  |  | 3 | 3 | 3 | 3 |

**S - Strong (3) M - Medium (2) L -Low**

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| **SEMESTER: II**  **PART: III** | **23UZOOEP2**  **Zoology Practical – II** | **Credit: 1**  **Hours: 2** |

Course Objectives:

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| 1 | To learn basic concepts relating to various physiological aspects of animals. |
| 2 | To comprehend the processes involved during development |
| 3 | To learn basic concepts of immunity and familiarize on immune organs. |
| 4 | To know the basic concepts of human genetics and patterns of inheritance |
| 5 | To learn about aspects of animal behaviour. |

**Practicals:**

1. Qualitative detection of excretory products (Ammonia, Urea, Uric acid).
2. Frog Egg, Blastula and Gastrula.
3. Demonstration of lymphoid organs.

4. Identification of ABO blood groups

5. Identification of human syndroms from karyotyping

5. Vital staining of chick blastoderm

7. Study of behavioural adaptations of animals

**Expected Course Outcomes**

On completion of this course, students will be able to:

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| --- | --- |
| 1 | Recall the parts and working of body organs |
| 2 | Analyse the different developmental stages |
| 3 | Analyse the functioning of body and immune systems |
| 4 | Analyse the different patterns of inheritance |
| 5 | Understand the different types of behaviour |

**Text Book(s)**

1 Arumugam N. (2013). Developmental Zoology, Saras Publication, Nagercoil, Tamilnadu, India.

2 Das S. (2020).Microbiology Practical Manual, CBS Publication, Delhi.

3 Jayasurya, Arumugam N, Dulsy Fatima. (2013). Practical Zoology Vol 3, Saras Publication, Nagercoil, Tamilnadu, India.

4 Singh HR and Neerajkumar. (2014). Animal Physiology and Biochemistry, Vishal Publishing Co. Jalandhar, Delhi.

**Outcome Maping**

|  |  |  |  |  |  |  |  |  |
| --- | --- | --- | --- | --- | --- | --- | --- | --- |
|  | **PO 1** | **PO 2** | **PO 3** | **PO 4** | **PO 5** | **PO 6** | **PO 7** | **PO 8** |
| **CO 1** | S | M |  | M | M | S |  | S |
| **CO 2** | M | S | M |  | S |  | M |  |
| **CO 3** | S | M |  | S |  | S |  | M |
| **CO 4** | S | S |  | S | S | M |  |  |
| **CO 5** | S | S | S |  |  |  | S | S |

**S-Strong M-Medium L-Low**

Non-major (NME) Electives offered to other Department

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| **SEMESTER: I**  **PART: III** | **23UCHEN16**  **ROLE OF CHEMISTRY IN DAILY LIFE** | **Credit: 2**  **Hours: 2** |

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| Objectives of the course | This course aims at providing an overall view of the   * importance of Chemistry in everyday life * chemistry of building materials and food * chemistry of Drugs and pharmaceuticals |
| Course  Outline | UNIT-I  General survey of chemicals used in everyday life. Air - components and their importance; photosynthetic reaction, air pollution, green - house effect and the impact on our life style. Water - Sources of water, qualities of potable water, soft and hard water, methods of removal of hardness-water pollution |
| Unit-II  Building materials - cement, ceramics, glass and refractories - definition, composition and application only. Plastics - polythene, PVC, bakelite, polyesters, melamine-formaldehyde resins -preparation and uses only. |
| UNIT-III  Food and Nutrition - Carbohydrates, Proteins, Fats - definition and their importance as food constituents – balanced diet – Calories minerals and vitamins (sources and their physiological importance). Cosmetics – tooth paste, face powder, soaps and detergents, shampoos, nail polish, perfumes - general formulation and preparations - possible hazards of cosmetic use. |
| UNIT-IV  Chemicals in food production – fertilizers - need, natural sources; urea, NPK fertilizers and super phosphate. Fuel – classification - solid, liquid and gaseous; nuclear fuel examples and uses. |
| UNIT-V  Pharmaceutical drugs - analgesics and antipyretics - paracetamol and aspirin. Colour chemicals - pigments and dyes - examples and applications.  Explosives - classification and examples. |
|  |  |
| Recommende  d Text | 1.Food chemistry, H. K. Chopra, P. S. Panesar, Narosa publishing house, 2010.   1. A textbook of pharmaceutical chemistry by Jayashree Ghosh, S Chand publishing, 2012. 2. S. Vaithyanathan, Text book of Ancillary Chemistry; Priya Publications, Karur, 2006. 3. B. K, Sharma, Industrial Chemistry; GOEL publishing house, Meerut, sixteenth edition, 2014.Introduction to forensic chemistry, Kelly M. Elkins, CRC Press Taylor & Francis Group, 2019. 4. Jayashree Ghosh, Fundamental Concepts of Applied Chemistry, S. Chand &Co.Publishers, second edition, 2006. |
| Reference  Books | 1. Randolph. Norris Shreve, Chemical Process Industries, McGraw-Hill, Texas, fourthedition, 1977. 2. W.A.Poucher,JosephA.Brink,Jr.Perfumes,Cosmetics and Soaps,Springer, 2000. 3. A.K.De,EnvironmentalChemistry,NewAge InternationalPublicCo.,1990. |
| Website and e-learning source | 1. <https://www.ncbi.nlm.nih.gov/pmc/articles/PMC7044178/> 2. <https://byjus.com/question-answer/name-the-element-which-is-important-component-of-ceramics-glass-and-cement-csialca-1/> 3. <https://kids.britannica.com/students/article/food-and-nutrition/274373> 4. https://study.com/academy/lesson/pharmaceutical-drugs-definition-types.html |
| Course Learning Outcomes (for Mapping with POs and PSOs)  On completion of the course the students should be able to  CO1: learn about the chemicals used in everyday life as well as air pollution and water pollution.  CO2: get knowledge on building materials cement, ceramics, glass and plastics, polythene, PVC bakelite, polyesters,  CO3: acquire information about Food and Nutrition. Carbohydrates, Proteins, Fats Also have an awareness about Cosmetics Tooth pastes, face powder, soaps and detergents.  CO4: discuss about the fertilizers like urea, NPK fertilizers and super phosphate. Fuel classification solid, liquid and gaseous; nuclear fuel - examples and uses  CO5: have an idea about the pharmaceutical drugs analgesics and antipyretics like paracetamol and aspirin and also about pigments and dyes and its applications. | |



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| --- | --- | --- | --- | --- | --- |
| CO /PO | PSO1 | PSO2 | PSO3 | PSO4 | PSO5 |
| CO1 | 3 | 3 | 3 | 3 | 3 |
| CO2 | 3 | 3 | 3 | 3 | 3 |
| CO3 | 3 | 3 | 3 | 3 | 3 |
| CO4 | 3 | 3 | 3 | 3 | 3 |
| CO5 | 3 | 3 | 3 | 3 | 3 |
| Weightage | 15 | 15 | 15 | 15 | 15 |
| Weighted percentage of  Course Contribution to  Pos | 3.0 | 3.0 | 3.0 | 3.0 | 3.0 |

Level of Correlation between PSO’s and CO’s

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| --- | --- | --- |
| **SEMESTER: II**  **PART: III** | **23UCHEN26**  **DAIRY CHEMISTRY** | **Credit: 2**  **Hours: 2** |

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| Objectives of the course | This course aims at providing an overall view of the   * chemistry of milk and milk products * processing of milk * preservation and formation of milk products. |
| Course Outline | UNIT I  Composition of Milk  Milk-definition-general composition of milk- constituents of milk - lipids, proteins, carbohydrates, vitamins and minerals - physical properties of milk - colour, odour, acidity, specific gravity, viscosity and conductivity -Factors affecting the composition of milk - adulterants, preservatives with neutralizer- examples and their detection- estimation of fat, acidity and total solids in milk. |
| Unit II  Processing of Milk  Microbiology of milk - destruction of micro - organisms in milk, physico – chemical changes taking place in milk due to processing - boiling, pasteurization – types of pasteurization -Bottle, Batch and HTST (High Temperature Short Time) – Vacuum pasteurization – Ultra High Temperature Pasteurization. |
| UNIT III  Major Milk Products  Cream - definition - composition - chemistry of creaming process - gravitational and centrifugal methods of separation of cream - estimation of fat in cream. Butter - definition -composition - theory of churning – desi butter - salted butter, estimation of acidity and moisture content in butter. Ghee - major constituents - common adulterants added to ghee and their detection - rancidity - definition - prevention - antioxidants and synergists - natural and synthetic. |
| UNIT IV  Special Milk  Standardised milk - definition - merits - reconstituted milk - definition - flow diagram of manufacture - Homogenised milk - flavoured milk - vitaminised milk - toned milk -Incitation milk - Vegetable toned milk - humanized milk - condensed milk - definition, composition and nutritive value. |
|  | UNIT V  Fermented and other Milk Products  Fermented milk products – fermentation of milk - definition, conditions, cultured milk - definition of culture - example, conditions - cultured cream, butter milk - Bulgarious milk -acidophilous milk – YoheerIndigeneous products- khoa and chhena definition - Ice cream -definition-percentage composition-types-ingredients-manufacture of ice–cream, stabilizers emulsifiersandtheirrole-milkpowder-definition-needformakingmilkpowderdryingprocess-types of drying. |